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Highly Enantioselective Friedel–Crafts Reaction of 4,7-Dihydroindoles with β , γ -Unsaturated α -Keto Esters by Chiral Brønsted Acids

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Dedicated to Prof. Xiyan Lu on the occasion of his 80th birthday.

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Abstract: A highly efficient Friedel–Crafts reaction of 4,7-dihydroindoles with β , γ -unsaturated α -keto esters by a chiral *N*-triflyl phosphoramide was realized, affording the 2-substituted 4,7-dihydroindoles with up to 98% *ee* for a wide range of substrates. The Friedel–Crafts alkylation together with a subsequent oxidation of the product with *p*-benzoquinone led to a 2-alkylated indole derivative in 98% *ee*.

Keywords: asymmetric catalysis; chiral Brønsted acids; conjugate addition; enantioselectivity; Friedel–Crafts reaction; indoles

Indoles are probably the most widely distributed heterocyclic compounds in nature and exist extensively as the structure core of biologically active natural products and pharmaceutical compounds.^[1] Therefore the synthesis of optically pure indole derivatives is one of the most intense research areas in organic synthesis.^[2] In this regard, the enantioselective Friedel-Crafts reaction of indole represents one of the most important reactions for the direct access to optically pure indoles.^[3] Although the enantioselective Friedel– Crafts reaction has attracted considerable interest and witnessed significant progress recently, [4] the highly enantioselective synthesis of 2-substituted indole derivatives is still a challenge since enantioselective Friedel-Crafts reactions of indole always lead to 3substituted indole derivatives.^[5]

Just recently, following the protocol of Saraçoğlu, ^[6] Evans and co-workers have realized the Lewis acid-catalyzed asymmetric Friedel–Crafts reaction of α , β -

unsaturated 2-acylimidazoles with 4,7-dihydroindoles.^[7] This elegant protocol provides an easy access to 2-substituted indole derivatives after a simple oxidation of the products generated therefrom. Similarly, Pedro and co-workers reported a catalytic Friedel-Crafts alkylation at the 2-position of indole with simple enones, giving moderate ees in the presence of a zirconium(IV)-BINOL complex.[8] Interesting as the optically pure 2-substituted indole derivatives are, their catalytic asymmetric synthesis is still rather limited. For instance, the alkylation of β , γ -unsaturated α keto esters at the 2-position of indole has not been reported yet although both the ketone and ester groups in the products would offer facile conversions to versatile functionalities.^[9] Taking advantage of Brønsted acid catalysis,[10-11] we recently realized the chiral phosphoric acid-catalyzed Friedel-Crafts reaction of 4,7-dihydroindoles with imines, affording the 2-indolylmethanamine derivatives after a subsequent oxidation. [11a] Moreover, Yamamoto and co-workers have recently introduced the chiral N-triflyl phosphoramide as a stronger acid compared with the chiral phosphoric acid, and the former enables the activation of carbonyl groups and highly extends the reaction type catalyzed by chiral Brønsted acids. [12] We envisaged that the chiral N-triflyl phosphoramide might be able to activate the β , γ -unsaturated α -keto ester to realize the Friedel-Crafts alkylation of indole at the 2-position (Figure 1). It should be noted that, very recently, Rueping and co-workers reported the chiral Ntriflyl phosphoramide-catalyzed enantioselective Friedel-Crafts alkylation of indole with β,γ -unsaturated α-keto esters with up to 92% ee. [12c] In this paper, we report a highly enantioselective Friedel-Crafts reaction of 4,7-dihydroindoles with β , γ -unsaturated α -keto esters catalyzed by the chiral N-triflyl phosphoramide,



Figure 1. Route to 2-substituted indoles *via* Friedel–Crafts alkylation of 4,7-dihydroindoles.

giving the 2-substituted indole derivatives with up to 98% ee.

We first examined the reaction between 4,7-dihydroindole **2a** and β , γ -unsaturated α -keto ester **3a** catalyzed by different chiral *N*-triflyl phosphoramides (see the Supporting Information for details). In the presence of 5 mol% of the chiral *N*-triflyl phosphoramides in toluene at $-78\,^{\circ}$ C, reactions of **3a** with 2 equivalents of **2a** all gave the desired product **4aa** smoothly together with 1,2-addition byproduct, and the best *ee* (89% *ee*) was obtained using catalyst **1** bearing the bulky 2,4,6-(*i*-Pr)₃C₆H₂ groups (entry 5, Table 1).

With 5 mol% of (S)-1 as the catalyst, reaction temperatures and solvents have been examined, and the results are summarized in Table 1. Interestingly, the reaction at -60 °C gave a higher ee (92%) with a rea-

sonable yield (84%) (entry 4, Table 1). Several common solvents such as dichloromethane (CH₂Cl₂), THF, diethyl ether (Et₂O), and EtOAc have been tested. All the solvents led to excellent enantiomeric excess values, and reaction in diethyl ether (Et₂O) gave the optimal result, 96% yield and 98% *ee* (entry 8, Table 1).

As listed in Table 2, the catalyst loading and the equivalents of 4,7-dihydroindole have been examined.

Table 2. Optimization of the reaction conditions.[a]

$$\begin{array}{c} & & & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & \\ & & \\ & \\ & & \\ & \\ & & \\ & \\ & \\ & & \\ & \\ & \\ & & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ &$$

Entry	y (equiv)	x (mol%)	Time (h)	Yield [%] ^[b]	ee [%] ^[c]
1	2.0	5	1.0	96	98
2	1.5	5	1.0	84	98
3	1.1	5	2.0	74	94
4	2.0	2	1.0	89	97
5	2.0	1	2.0	85	94

- [a] Reaction conditions: y equiv of 2a, x mol% 1, -60°C, 0.20 mol/L of 3a in ether.
- [b] Isolated yields.
- [c] Determined by chiral HPLC analysis (Chiralcel OD-H column).

Table 1. Optimization of the reaction conditions for enantioselective Friedel-Crafts reactions.^[a]

Entry	Temperature [°C]	Solvent	Time [h]	Yield [%] ^[b]	ee [%] ^[c]
1	25	toluene	0.25	52	64
2	0	toluene	1.0	80	84
3	-40	toluene	1.0	77	90
4	-60	toluene	1.5	84	92
5	-78	toluene	2.5	87	89
6	-60	dichloromethane (CH ₂ Cl ₂)	1.0	70	87
7	-60	tetrahydrofuran (THF)	8.0	48	86
8	-60	diethyl ether (Et ₂ O)	1.0	96	98
9	-60	EtOAc	1.0	80	90

[[]a] Reaction conditions: 2.0 equiv. of 2a, 5 mol% of 1, 0.20 mol/L of 3a.

[[]b] Isolated yields.

[[]c] Determined by chiral HPLC analysis (Chiralcel OD-H column).

With 1.5 or 1.1 equivalents of dihydroindole **2a**, the reaction proceeded smoothly to give the alkylation product in 98% and 94% *ee*, respectively. Gratifyingly, even with 1 mol% of the catalyst, alkylation product **4aa** was also obtained in 85% yield with 94% *ee* (entry 5, Table 2).

Under the above optimized reaction conditions (2.0 equivalents of 4,7-dihydroindole, 5 mol% of 1, -60°C, and Et₂O as the solvent), Friedel–Crafts reactions between a wide range of β , γ -unsaturated α -keto esters 3 and substituted 4,7-dihydroindoles have been carried out to test the generality of the reaction. The results are summarized in Table 3.

The chiral phosphoramide-catalyzed Friedel-Crafts reaction of 4,7-dihydroindoles with β , γ -unsaturated α keto esters was found to be general with keto esters bearing different substituents. Several substituted keto esters **3b-g**, containing electron-donating groups at the para- or meta-position of the phenyl ring, have been tested in the reaction with 4,7-dihydroindole 2a. In all cases, moderate to high yields and excellent enantioselectivities could be achieved for the desired alkylation products (59 to 82% yield, 87 to 97% ee, entries 2-7, Table 3). The chemistry went also well with substituted keto esters 3h-l, containing electronwithdrawing groups at the para- or meta-position of the phenyl ring, and the desired alkylation products were obtained in 66 to 85% yield with 91 to 97% ee (entries 8–12, Table 3). When γ-heteroaryl-substituted α-keto esters **3m**, **n** were used, the reaction also gave excellent results, 89% yield with 96% *ee* (2-furyl), and 75% yield with 98% *ee* (2-thienyl), respectively (entries 13 and 14, Table 3). 5-Fluoro-4,7-dihydroindole **2b** has been tested in the reaction with keto ester **3a**, the alkylation product **4ba** was obtained in 84% yield with 96% *ee* (entry 15, Table 3).

Interestingly, when 5-methoxy-4,7-dihydroindole **2c** was tested in the reaction with keto ester **3a**, product **5** was obtained in 53% yield with 94% *ee* due to the hydrolysis of the enol methyl ether (Scheme 1).

To demonstrate the suitability of the current methodology for the synthesis of 2-substituted indole derivatives, the oxidation of the 2-substituted 4,7-dihydroindole derivative has been tested. As shown in Scheme 2, after a quick work-up of the Friedel-Crafts

Scheme 1. Friedel–Crafts reaction of 5-methoxy-4,7-dihydroindole **2c** with keto ester **3a**.

Table 3. Enantioselective Friedel–Crafts reaction of 4,7-dihydroindoles with β , γ -unsaturated α -keto esters. [a]

R¹ O CO₂Et (S)-1 (5 mol%) R¹
$$= F$$
 $= F$ $= F$

Entry	2	3, R ²	Time (h)	4 , Yield [%] ^[b]	ee [%] ^[c]
1	2a	3a , C ₆ H ₅	1.0	4aa , 96	98
2	2a	3b , 4-Me- C_6H_4	12	4ab , 59	87
3	2a	3c , 3 -Me-C ₆ H ₄	2.5	4ac , 82	97
4	2a	3d , 3 -MeO-C ₆ H ₄	1.5	4ad , 80	96
5	2a	3e , 4 -MeO-C ₆ H ₄	1.0	4ae , 75	97
6	2a	3f , piperonyl	1.0	4af , 82	97
7	2a	3g , 2-naphthyl	30	4ag , 74	93
8	2a	3h , $3-NO_2-C_6H_4$	24	4ah , 66	93
9	2a	3i, 4 -Cl-C ₆ H ₄	2.5	4ai , 83	91
10	2a	3j, 4-Br-C ₆ H ₄	1.0	4aj , 85	96
11	2a	3k , 4-F-C ₆ H ₄	3.0	4ak , 80	91
12	2a	31 , 3-Br- C_6H_4	2.0	4al , 85	96
13	2a	3m , 2-furyl	1.0	4am , 89	96
14	2a	3n , 2-thienyl	2.5	4an , 75	98
15	2b	$3a, C_6H_5$	12	4ba , 84	96

[[]a] Reaction conditions: 2.0 equiv. of **2**, 5 mol% **1**, -60 °C, 0.20 mol/L of **3** in ether.

[[]b] Isolated yields.

[[]c] Determined by chiral HPLC analysis.

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Scheme 2. Friedel–Crafts reaction of **2a** with keto ester **3a** and *p*-benzoquinone oxidation.

reaction of 4,7-dihydroindole **2a** with keto ester **3a**, the reaction mixture was subjected to 2 equivalents of *p*-benzoquinone in acetonitrile. The desired 2-indolyl compound **6** was obtained smoothly in an overall 59% yield with 98% *ee*, which indicated the perfect retention of the stereochemistry during the oxidation process (entry 1, Table 3).

In summary, we have developed the enantioselective Friedel–Crafts reaction of 4,7-dihydroindoles with β,γ -unsaturated α -keto esters by utilizing chiral *N*-triflyl phosphoramide **1** as an efficient catalyst. The reaction features a metal-free approach, high efficiency of the catalyst, mild reaction conditions, high yields, and excellent enantioselectivities, providing a practical method to synthesize highly enantiopure 2-substituted 4,7-dihydroindole and 2-indole derivatives.

Experimental Section

General Procedure for the Catalytic Asymmetric Friedel-Crafts Reaction

In a dry Schlenk tube, β,γ -unsaturated α -keto ester **3** (0.20 mmol) and *N*-triflyl phosphoramide **1** (8.8 mg, 0.010 mmol) were dissolved in Et₂O (1 mL) under argon. The solution was stirred for 5 min at room temperature and then for another 5 min at $-60\,^{\circ}$ C. Subsequently, *N*-methyl-4,7-dihydroindole **2** (0.40 mmol) was added in one portion at $-60\,^{\circ}$ C. After the reaction was complete (monitored by TLC), saturated aqueous NaHCO₃ (3 mL) was added to quench the reaction. The mixture was allowed to warm to room temperature and was then extracted with CH₂Cl₂. The organic layer was separated and dried over anhydrous Na₂SO₄. The solvents were removed under reduced pressure and the residue was purified by flash chromatography (ethyl acetate/petroleum ether = 1/5–1/20) to afford the product.

Supporting Information

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Full experimental details with characterization data are available as Supporting information.

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References

- a) J. Bosch, M.-L. Bennasar, Synlett 1995, 587-596;
 b) D. J. Faulkner, Nat. Prod. Rep. 2002, 19, 1-48;
 c) S. Agarwal, S. Caemmerer, S. Filali, W. Froehner, J. Knoell, M. P. Krahl, K. R. Reddy, H.-J. Knoelker Curr. Org. Chem. 2005, 9, 1601-1614;
 d) S. E. O'Connor;
 J. Maresh, Nat. Prod. Rep. 2006, 23, 532-547.
- [2] For reviews: a) Y. Wang, K. Ding, L. Dai, Chemtracts 2001, 14, 610-615; b) M. Bandini, A. Melloni, A. Umani-Ronchi, Angew. Chem. 2004, 116, 560-566; Angew. Chem. Int. Ed. 2004, 43, 550-556; c) M. Bandini, A. Melloni, S. Tommasi, A. Umani-Ronchi, Synlett 2005, 1199-1222.
- [3] For reviews: a) G. A. Olah, in: Friedel-Crafts and Related Reactions, Wiley, New York, 1963; b) G. A. Olah, in: Friedel-Crafts Chemistry, Wiley, New York, 1973; c) G. A. Olah, R. Krishnamurti, G. K. S. Prakash, in: Comprehensive Organic Synthesis, Vol. 3, (Eds.: B. M. Trost, I. Fleming), Pergamon, Oxford, 1991, p 293; d) T. B. Poulsen, K. A. Jørgensen, Chem. Rev. 2008, 108, 2903-2915.
- [4] For selected examples: a) N. Gathergood, W. Zhuang, K. A. Jørgensen, J. Am. Chem. Soc. 2000, 122, 12517-12522; b) J. F. Austin, D. W. C. MacMillan, J. Am. Chem. Soc. 2002, 124, 1172-1173; c) J. Zhou, Y. Tang, J. Am. Chem. Soc. 2002, 124, 9030-9031; d) D. A. Evans, K. A. Scheidt, K. R. Fandrick, H. W. Lam, J. Wu, J. Am. Chem. Soc. 2003, 125, 10780-10781; e) Y. Yuan, X. Wang, X. Li, K. Ding, J. Org. Chem. 2004, 69, 146-149; f) S. Shirakawa, R. Berger, J. L. Leighton, J. Am. Chem. Soc. 2005, 127, 2858-2859; g) C. Palomo, M. Oiarbide, B. G. Kardak, J. M. García, A. Linden, J. Am. Chem. Soc. 2005, 127, 4154-4155; h) D. A. Evans, K. R. Fandrick, H.-J. Song, J. Am. Chem. Soc. 2005, 127, 8942-8943; i) Y.-X. Jia, J.-H. Xie, H.-F. Duan, L.-X. Wang, Q.-L. Zhou, Org. Lett. 2006, 8, 1621–1624; j) S.-F. Lu, D.-M. Du, J. Xu, Org. Lett. 2006, 8, 2115-2118; k) H. Li, Y.-Q. Wang, L. Deng, Org. Lett. 2006, 8, 4063-4065; l) J.-L. Zhao, L. Liu, Y. Sui, Y.-L. Liu, D. Wang, Y.-J. Chen, Org. Lett. 2006, 8, 6127-6130; m) W. Zhou, L.-W. Xu, L. Lyi, L. Yang, C.-G. Xia, Eur. J. Org. Chem. 2006, 23, 5225-5227; n) W. Chen, W. Du, L. Yue, R. Li, Y. Wu, L.-S. Ding, Y.-C. Chen, Org. Biomol. Chem. 2007, 5, 816-821; o) G. Bartoli, M. Bosco, A. Carlone, F. Pesciaioli, L. Sambri, P. Melchiorre, Org. Lett. 2007, 9, 1403-1405; p) C.-F. Li, H. Liu, J. Liao, Y.-J. Cao, X.-P. Liu, W.-J. Xiao, Org. Lett. 2007, 9, 1847–1850; q) H. Yang, Y.-T. Hong, S. Kim, Org. Lett. 2007, 9, 2281-2284; r) G. Blay, I. Fernández, J. R. Pedro, C. Vila, Org. Lett. 2007, 9, 2601-2604; s) H.-M. Dong, H.-H. Lu, L.-Q. Lu, C.-B. Chen, W.-J. Xiao, Adv. Synth. Catal. 2007, 349, 1597-1603; t) G.-W. Zhang, L. Wang, J. Nie, J.-A. Ma, Adv. Synth. Catal.

- **2008**, *350*, 1457–1463; u) W.-B. Liu, H. He, L.-X. Dai, S.-L. You, *Org. Lett.* **2008**, *10*, 1815–1818.
- [5] For exceptional examples: a) M. S. Taylor, E. N. Jacobsen, J. Am. Chem. Soc. 2004, 126, 10558-10559; b) J. Seayad, A. M. Seayad, B. List, J. Am. Chem. Soc. 2006, 128, 1086-1087; c) I. T. Raheem, P. S. Thiara, E. A. Peterson, E. N. Jacobsen, J. Am. Chem. Soc. 2007, 129, 13404-13405; d) S. Lee, D. W. C. MacMillan, J. Am. Chem. Soc. 2007, 129, 15438-15439; e) M. J. Wanner, R. N. S. vander Haas, K. R. deCuba, J. H. van Maarseveen, H. Hiemstra, Angew. Chem. 2007, 119, 7629-7631; Angew. Chem. Int. Ed. 2007, 46, 7485-7487.
- [6] a) H. Çavdar, N. Saraçoğlu, Tetrahedron 2005, 61, 2401–2405; b) H. Çavdar, N. Saraçoğlu, J. Org. Chem. 2006, 71, 7793–7799.
- [7] a) D. A. Evans, K. R. Fandrick, Org. Lett. 2006, 8, 2249–2252; b) D. A. Evans, K. R. Fandrick, H.-J. Song, K. A. Scheidt, R. Xu, J. Am. Chem. Soc. 2007, 129, 10029–10041.
- [8] G. Blay, I. Fernández, J. R. Pedro, C. Vila, *Tetrahedron Lett.* 2007, 48, 6731–6734.
- [9] For the Friedel–Crafts reaction of β,γ-unsaturated α-keto esters: a) K. B. Jensen, J. Thorhauge, R. G. Hazell, K. A. Jøgensen, Angew. Chem. 2001, 113, 164–167; Angew. Chem. Int. Ed. 2001, 40, 160–163; b) K. A. Jøgensen, Synthesis 2003, 1117–1125; c) H. L. Van Lingen, W. Zhuang, T. Hansen, F. P. J. T. Rutjes, K. A. Jøgensen, Org. & Biomol. Chem. 2003, 1, 1953–1958;

- d) C. Üenaleroğlu, B. Temelli, A. S. Demir, *Synthesis* **2004**, 2574–2578; e) M. P. A. Lyle, N. D. Draper, P. D. Wilson, *Org. Lett.* **2005**, *7*, 901–904.
- [10] Reviews on chiral phosphoric acid catalysis: a) M. S. Taylor, E. N. Jacobsen, Angew. Chem. 2006, 118, 1550–1573; Angew. Chem. Int. Ed. 2006, 45, 1520–1543; b) T. Akiyama, J. Itoh, K. Fuchibe, Adv. Synth. Catal. 2006, 348, 999–1010; c) S. J. Connon, Angew. Chem. 2006, 118, 4013–4016; Angew. Chem. Int. Ed. 2006, 45, 3909–3912; d) T. Akiyama, Chem. Rev. 2007, 107, 5744–5758.
- [11] a) Q. Kang, X.-J. Zheng, S.-L. You, Chem. Eur. J. 2008, 14, 3539-3542; b) Q. Kang, Z.-A. Zhao, S.-L. You, Org. Lett. 2008, 10, 2031-2034; c) Q. Kang, Z.-A. Zhao, S.-L. You, J. Am. Chem. Soc. 2007, 129, 1484-1485; d) Q. Kang, Z.-A. Zhao, S.-L. You, Adv. Synth. Catal. 2007, 349, 1657-1660.
- [12] a) D. Nakashima, H. Yamamoto, J. Am. Chem. Soc. 2006, 128, 9626-9627; b) M. Rueping, W. Ieawsuwan, A. P. Antonchick, B. J. Nachtsheim, Angew. Chem. 2007, 119, 2143-2146; Angew. Chem. Int. Ed. 2007, 46, 2097-2100; c) M. Rueping, B. J. Nachtsheim, S. A. Moreth, M. Bolte, Angew. Chem. 2008, 120, 603-606; Angew. Chem. Int. Ed. 2008, 47, 593-596; d) P. Jiao, D. Nakashima, H. Yamamoto, Angew. Chem. 2008, 120, 2445-2447; Angew. Chem. Int. Ed. 2008, 47, 2411-2413; e) D. Enders, A. A. Narine, F. Toulgoat, T. Bisschops, Angew. Chem. 2008, 120, 5744-5748; Angew. Chem. Int. Ed. 2008, 47, 5661-5665.